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Analytical Method Development and Validation of Favipiravir in Bulk and Tablet Formulation



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ABSTRACT

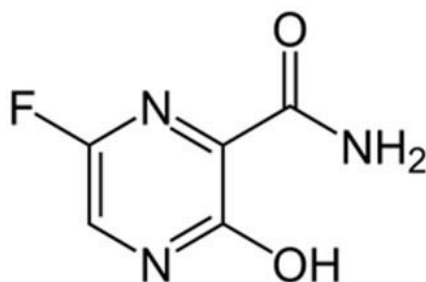
Favipiravir is an antiviral that is active against many RNA viruses, & also used in COVID-19. A simple, fast & reliable UV-spectroscopic method was developed for the estimation of Favipiravir in a bulk and pharmaceutical dosage form. It is a pyrazine carboxamide derivative. The solution of standard & sample was prepared in ethanol. The λ_{\max} was found to be 323 nm. Pure drug concentration were prepared in the range of 100 μ g/ml and the linear regression analysis data showed a good linear relationship with an R² value of 0.9994. The calibration graphs constructed at their wavelength of determination were found to be linear for UV and derivative spectrophotometric methods. The developed method was validated as per ICH guidelines by concerning accuracy, precision, linearity, limit of detection (LOD), limit of quantification (LOQ), & ruggedness. The LOD, LOQ & % RSD indicate that the developed method is suitable for the analysis of commercial samples.



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INTRODUCTION

Favipiravir, is one among potential drugs that may possibly be used in the management of COVID-19 infection. It is a pyrazine carboxamide derivative, that showed effective antiviral activity against a variety of RNA viruses including influenza A virus, adenovirus, and SARS Corona virus [1]. Favipiravir (6-fluoro-3-hydroxypyrazine-2-carboxamide) is a pyrazine analog. The mechanism of action is linked to transcription inhibition and viral gene replication which finally prevents the synthesis of viral RNA inside infected cells. [2] Literature search reveals that there are fewer method reported for the determination and quantification of Favipiravir and there is less research work done on this antiviral drug. No official pharmacopoeial method found for this particular antiviral drug. In the view of this criterion, the main aim and objective of this study is to develop and validate a sensitive method for the estimation of Favipiravir with applicability of Favipiravir in routine analysis [3].



The chemical structure of Favipiravir [4]

The present work was carried out with the view of establishing a simple, rapid, accurate, economic, precise, and robust UV method for the estimation of favipiravir in bulk and capsule dosage form, using methanol as the solvent. [2]

MATERIALS AND METHODS

Chemicals and Equipment

The pharmaceutical grade of Favipiravir was obtained from Swapnroop Reaserch Private Limited, Aurangabad. The ethanol is used as solvent. The tablet formulation of Favipiravir was bought from local pharmacy. Shimadzu UV 1800 double beam UV-visible spectrophotometer was used for UV analysis and for weighing purpose digital weighing balance was used.

Preparation of Standard solution

Stock solution of the drug 100 $\mu\text{g/ml}$ is prepared by dissolving 5 mg each drug in 50 ml volumetric flask and the volume is make up by ethanol.

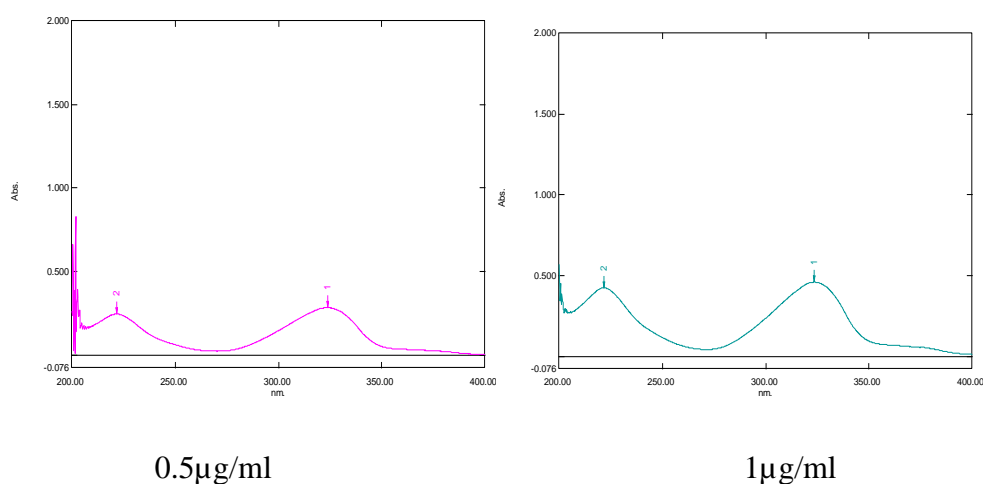
Selection of Analytical Wavelength

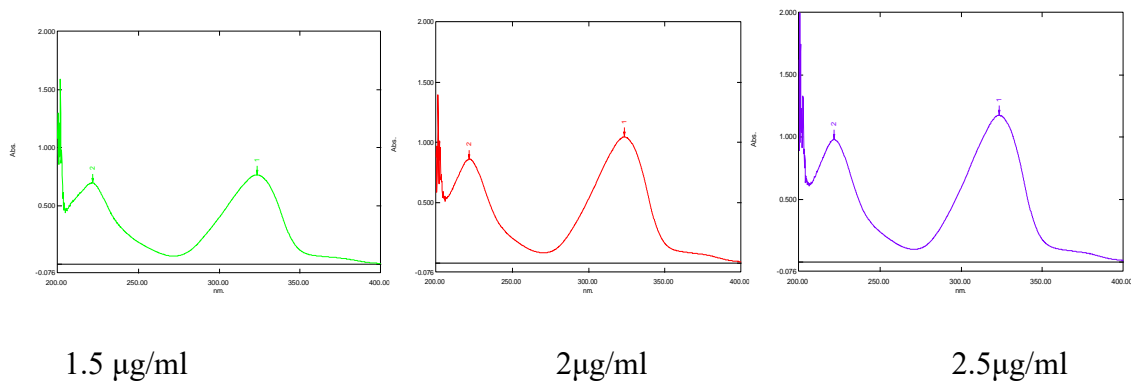
By appropriate dilution of standard solution of FVP 2 $\mu\text{g/ml}$ of each was prepared and scanned. Following peak were observed.

Drug	Lamda Max At Zero Order
FVP	323.08nm

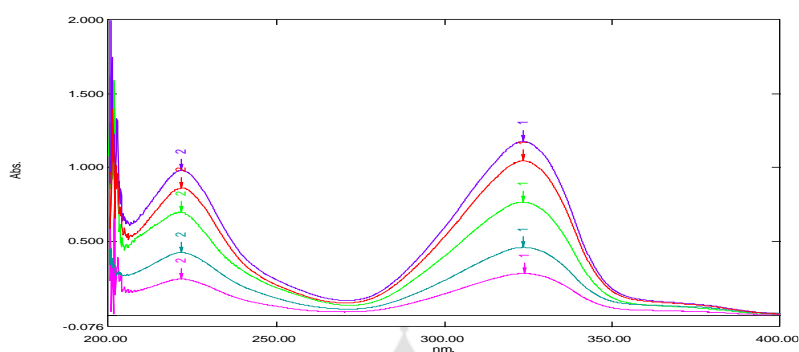
Zero order derivative spectroscopy

The derivative process provides two general advantages: first, an effective enhancement of resolution, which can be useful to separate two or more components with overlapping spectra; second, discrimination in favor of the sharpest features of a spectrum, used to eliminate interferences by broad band constituents. Five dilutions of FVP tab dilutions are prepared and there spectra are taken which are as follows:





Overlain of zero order spectroscopy spectra



First order derivative spectroscopy

The derivative spectrophotometric method is one of the advanced modern spectrophotometric techniques that offer a useful means for extracting both qualitative and quantitative information from the spectra composed of overlapped bands.

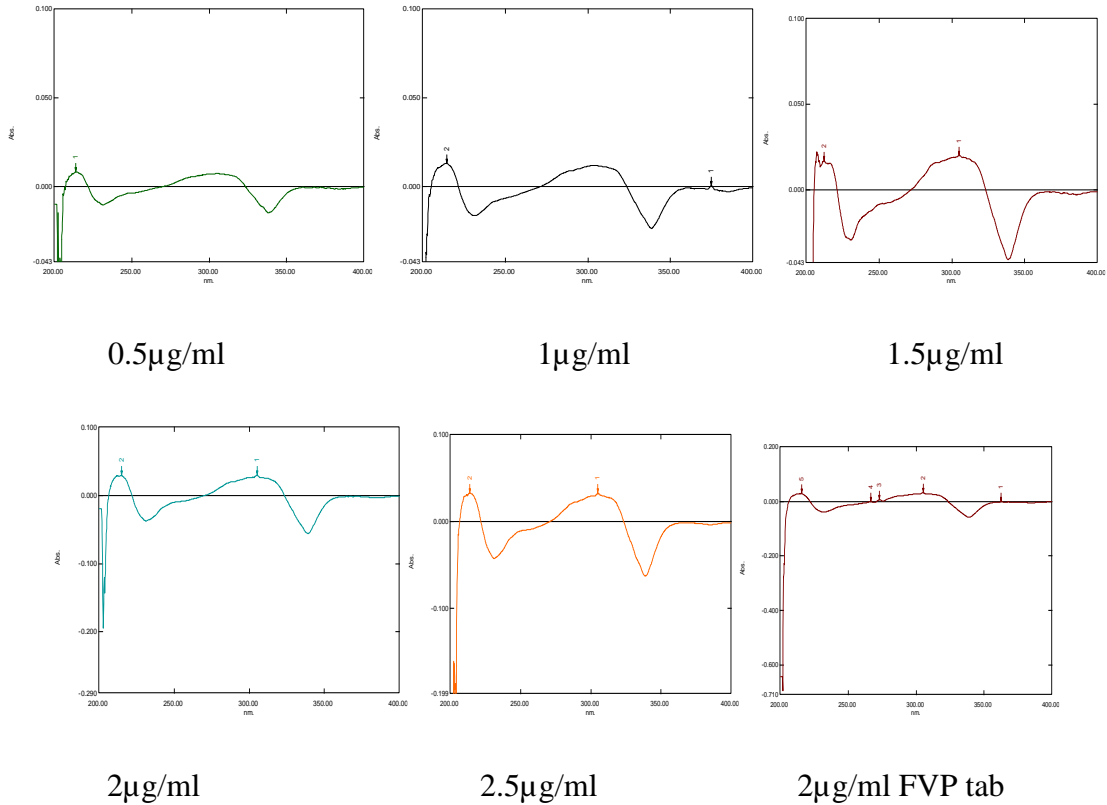
Solvent Used: Ethanol

Stock Solution:

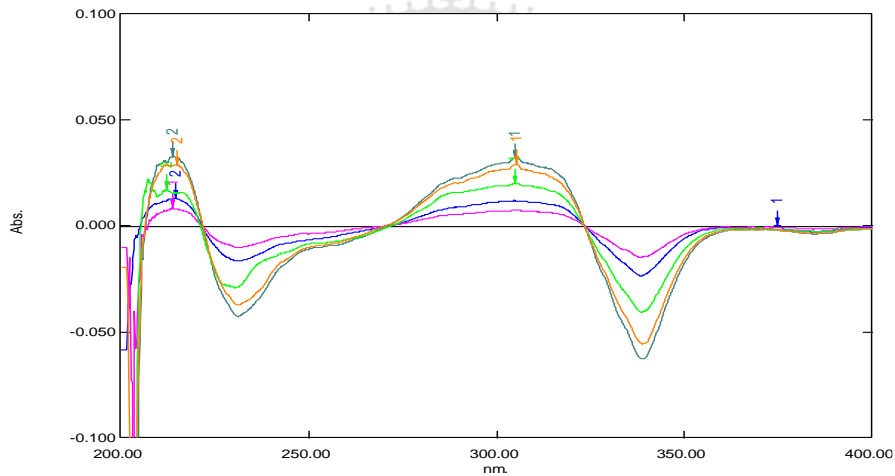
Stock solution of the drug 100 µg/ml is prepared by dissolving 5 mg each drug in 50 ml volumetric flask and the volume is make up by ethanol.

Selection of Analytical Wavelength

By appropriate dilution of standard solution of FVP 2 µg/ml of each were prepared and scanned. The λ_{\max} was found at 323nm. Five diutions are prepared and there spectra are taken which are as follows-

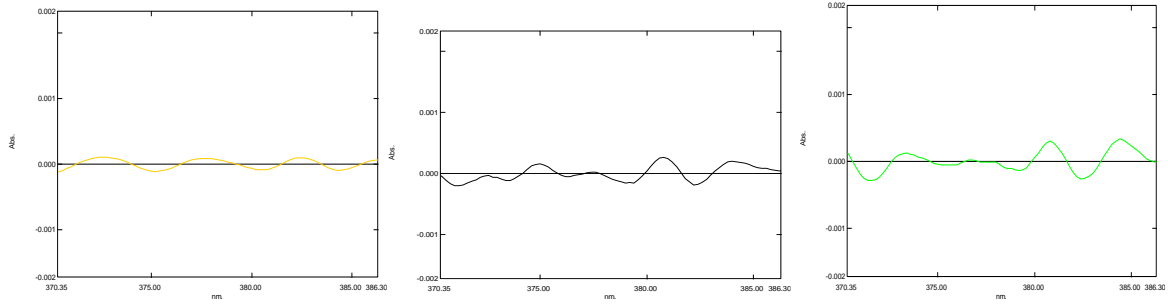


Overlain of first order spectroscopy spectra



Second order derivative spectroscopy:

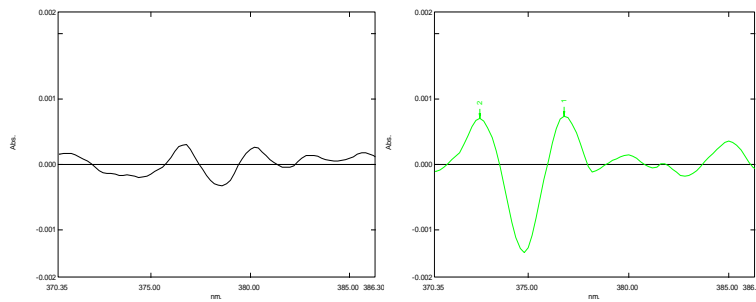
Second derivative spectroscopy is a technique which enhances the separation of overlapping peaks. The second derivative spectroscopy method requires the use of a UV/Visible scanning spectrophotometer. Five dilutions are prepared and their spectra are taken which are as follows-



0.5µg/ml

1µg/ml

1.5µg/ml

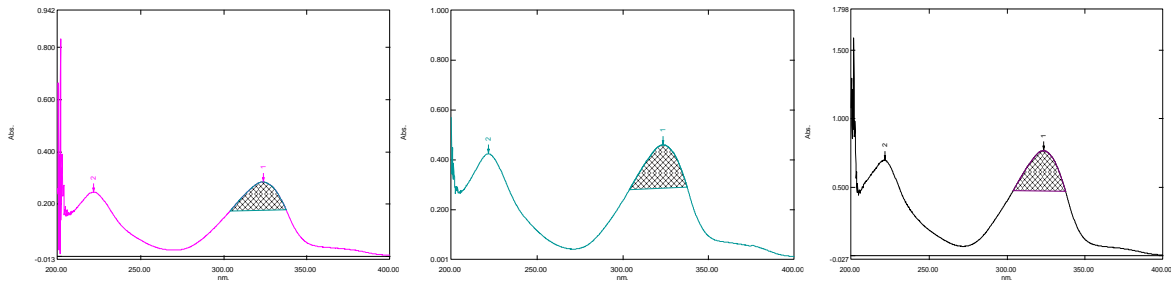


2µg/ml

2.5µg/ml

Area undercurve method

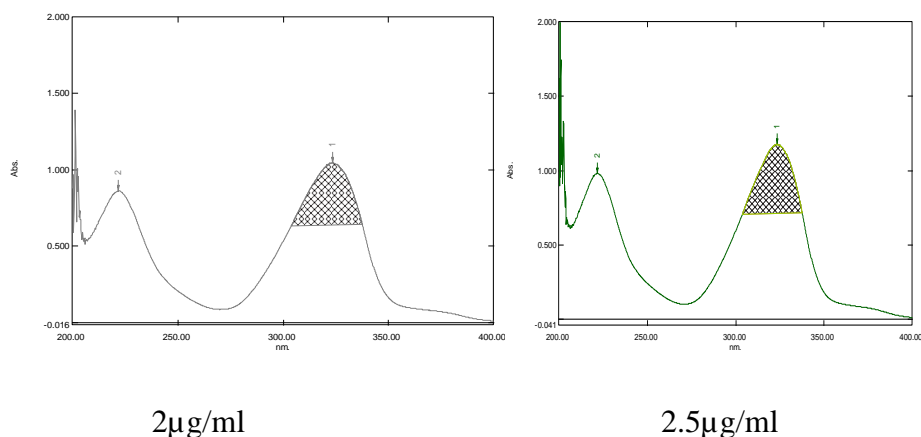
The AUC (area under curve) method is applicable where there is no sharp peak or when broad spectra are obtained. It involves the calculation of integrated value of absorbance with respect to the wavelength between the two selected wavelengths λ_1 and λ_2 . Area calculation processing item calculates the area bound by the curve and the horizontal axis. The horizontal axis is selected by entering the wavelength range over which area should be calculated. Five dilutions & FVP tab dilutions are prepared and there spectra are taken which are as follows-



0.5µg/ml

1µg/ml

1.5µg/ml



Method validation

Accuracy

Accuracy is a measure of the closeness of the experimental value to the actual amount of the substance in the matrix. It is one of the most critical parameter in method validation. It confirms the suitability of method to the greatest extent and hence method developer must design suitable extraction procedure to assure accurate quantification of analyte in presence of sample matrix.



Precision

Precision measures of how close individual measurements are to each other. It is the term used to see how reliable and consistent the measurement of the experiment is. Precision of a method is the degree of agreement among individual test results when the procedure is applied repeatedly to multiple samplings. It is measured by injecting a series of standards or analyzing series of samples from multiple samplings from a homogeneous lot.

Linearity

The linearity of an analytical procedure is its ability (within a given range) to obtain test results which are directly proportional to the concentration (amount) of analyte in the sample. Linearity is measuring the linear response of the method. Linearity studies are important because they define the range of the method within which the results are obtained accurately and precisely.

Limit of detection (LOD) and limit of quantification (LOQ)

Calculated based on the standard deviation of the response (S_y) of the curve and the slope of the calibration curve (S) at levels.

$$\text{Limit of detection (Lod)} = 3.3 * \text{standard deviation} / \text{slope}$$

$$\text{Limit of quantification (Loq)} = 10 * \text{standard deviation} / \text{slope}$$

Ruggedness

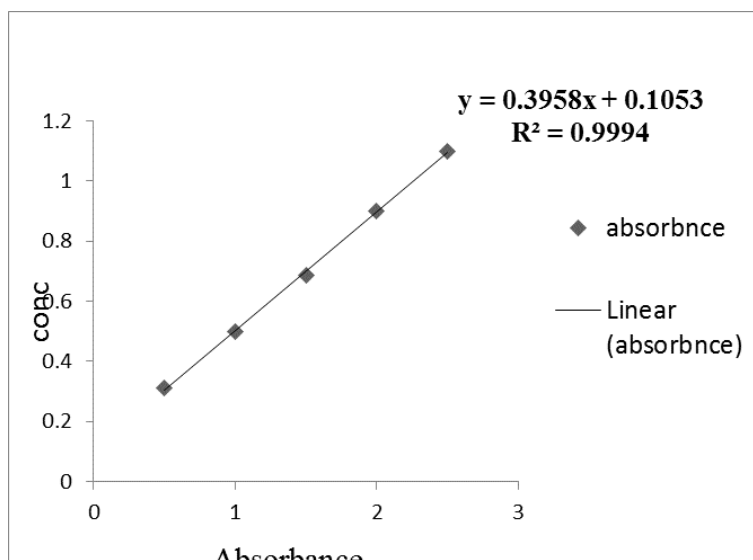
Ruggedness is normally expressed as the lack of influence on the test results of operational and environmental variables of the analytical method. It is a measure of reproducibility of test results under the variation in conditions normally expected from laboratory to laboratory & from analyst to analyst. The purpose of a ruggedness test is to find the factors that strongly influence measurement results, and to determine how closely one needs to control these factors.

RESULTS AND DISCUSSION

A calibration graph of concentration versus absorbance was plotted. The drug followed the Beers Lamberts law in the concentration range of 0.5-2.5 μ g/ml. Favipiravir gives λ_{max} at 323nm. The linearity graphs for spectroscopic methods are as shown in following graphs, the sensitivity of the method was assessed by determining LOD & LOQ which is found to be 0.3625 and 1.099 μ g/ml. Accuracy is performed with level of percent recovery 80,100 & 120.

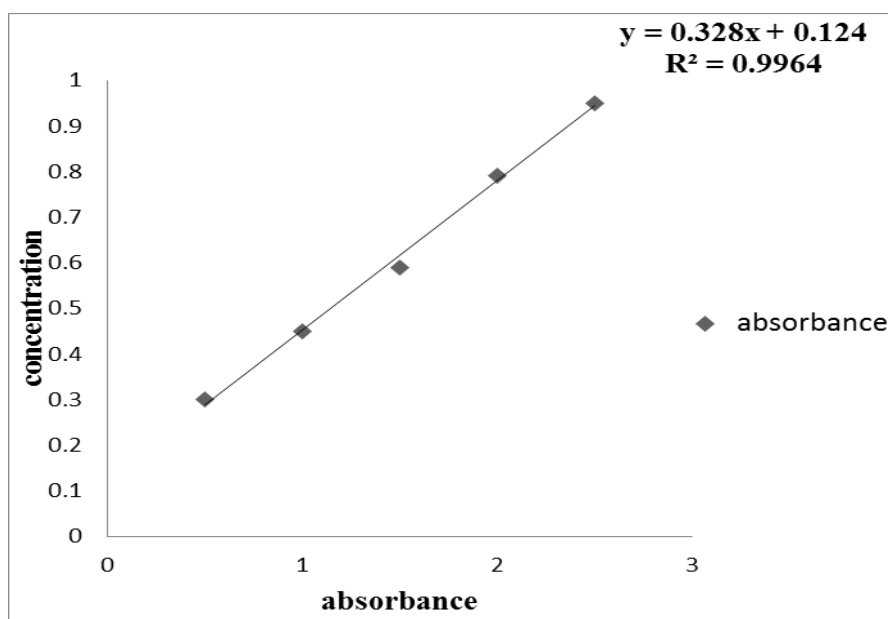
Table No. 1: Results of Linearity

Concentration	Absorbance
0.5	0.311
1	0.498
1.5	0.687
2	0.899
2.5	1.1
Regg. Equ.	$y=0.395+0.105$
R2	0.999



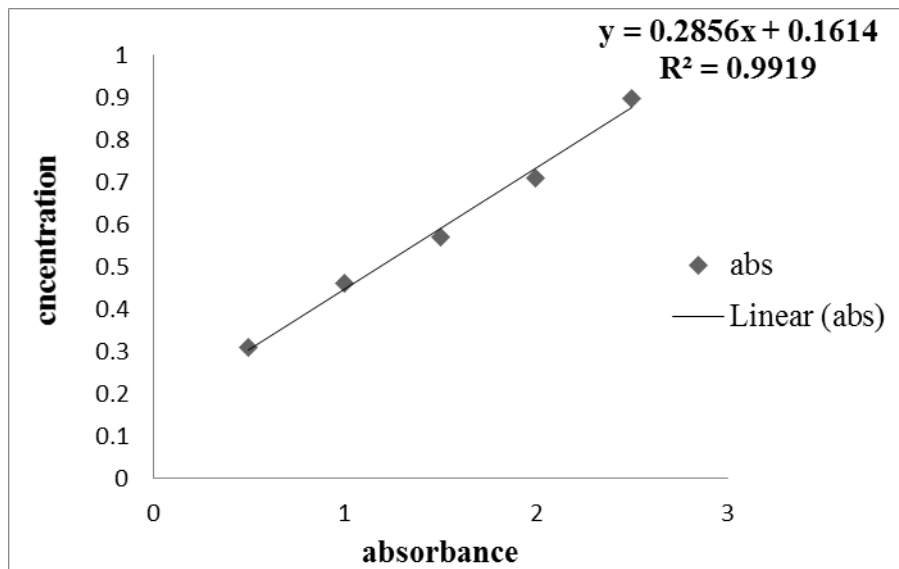
ZERO ORDER

Concentration	Absorbance
0.5	0.3
1	0.45
1.5	0.59
2	0.79
2.5	0.95
Regg. Equ.	$y=0.328x+0.124$
R2	0.996



FIRST ORDER

Concentration	Absorbance
0.5	0.31
1	0.46
1.5	0.57
2	0.71
2.5	0.899
Regg. Equ.	$y=0.285x+0.161$
R2	0.991



SECOND ORDER

	Zero order			First order			Second order		
Level of percent recovery	80	100	120	80	100	120	80	100	120
Amount present (Label claim)	200	200	200	200	200	200	200	200	200
Amount of standard added (mg)	160	200	240	160	200	240	160	200	240
Total amount recovered	359.1	400.3	438.2	358.2	399.1	438.5	359.7	400.2	439.6
% recovery	99.43	100.15	99.25	98.87	99.55	99.37	99.81	100.1	99.83
% mean	99.61			99.26			99.91		

Table No. 2: Result of Accuracy

Intraday			
Sr no.	Concentration (µg/ml)	Time	% recovery
1	2	Morning	99.43
2	2	Afternoon	98.87
3	2	Evening	99.25

Table No. 3: Result of Precision

Interday			
Sr no.	Concentration (µg/ml)	Day	% recovery
1	2	Day 1	99.10
2	2	Day 2	98.88
3	2	Day 3	98.59

Table No. 4: Result of Ruggedness

Sr no.	Concentration	Analyst 1 % recovery	Analyst 2 % recovery
1	2	99.87	99.44
2	2	99.25	99.84
3	2	99.89	99.51
4	2	99.79	99.20
5	2	99.58	99.37

CONCLUSION

No derivative spectroscopic methods have been described for the determination of Favipiravir. Therefore simple, fast, and reliable derivative spectroscopic methods were developed for the Favipiravir. The developed method can be concluded as accurate, sensitive and precise and can be easily applied to the pharmaceutical formulations.

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