



A Series of Novel 4-(Benzo[D]Oxazol) Piperazine Derivatives as Potent Anti-Fungal Agents: Design, Synthesis, Biological Evaluation and Molecular Docking Studies

Ms. Mona Patel¹, Ms. Jaychandrika Vasava², Dr. Ojas Patel*

¹Assistant Professor, Department of Quality assurance and Pharmaceutical Chemistry, Faculty of Pharmacy, SSSRGI, Vadasma, Mehsana- 382 705, Gujarat, India

²Department of Pharmaceutical Chemistry, Faculty of Pharmacy, SSSRGI, Vadasma, Mehsana- 382 705, Gujarat, India

*Professor, Department of Pharmaceutical Chemistry, Faculty of Pharmacy, SSSRGI, Vadasma, Mehsana- 382 705, Gujarat, India

Received: 19 February 2026

Revised: 28 February 2026

Accepted: 20 March 2026

ABSTRACT:

A series of 4-(benzo[d]oxazol) piperazine derivatives was designed and synthesized in excellent yield. All newly synthesized compound(s) were characterized by their infrared, proton nuclear magnetic resonance and mass spectral analysis. In order to evaluate their biological activity, all the compounds were screened for their antifungal activity against *Candida tropicalis* and *Aspergillus niger* employing a Agar well diffusion method. Furthermore, the experimental results were supported by molecular docking study. The results revealed that all newly synthesized compounds were exhibited moderate to potent antifungal activity. In case of potent antifungal activity, compound 1 gives prominent activity with comparison of all synthesized compounds. Significantly, we have designed, synthesized and characterized an interesting and biologically important series of 4-(benzo[d]oxazol) piperazine bridged analogues of Benzisoxazole derivatives. Most of the synthesized compounds were found more active as comparison with each other as manifested by theoretical as well as experimental results.

Key-words: Benzisoxazole, Antifungal, Piperazine, molecular docking.

INTRODUCTION: ¹⁻¹⁸

Fungal infections are caused by microscopic organisms that can invade the epithelial tissue. The fungal kingdom includes yeasts, molds, rusts and mushrooms.¹⁻³ Most fungi are beneficial and are involved in biodegradation; however, a few can cause opportunistic infections if they are introduced into the skin through wounds, or into the lungs and nasal passages if inhaled.⁴⁻⁵

Benzisoxazole is an aromatic organic compound with a molecular formula C_7H_5NO containing a benzene-fused isoxazole ring structure. It is a novel antiepileptic drug and is effective for the treatment of partial seizures. 1, 2-Benzisoxazole is a potential substrates of rabbit liver aldehyde oxidase. The supposed mechanism of action of benzisoxazole derived atypical antipsychotics is classified into serotonergic, dopaminergic and combined regulatory effects. Since the above antipsychotics mainly exhibit antagonism to D2/5HT2A/H1/ α adrenergic systems. It has a wide range of biological activities such as antimicrobial, anticonvulsant, antitumor, anti-psychotic, antithrombotic, analgesic activities.⁶⁻¹⁷

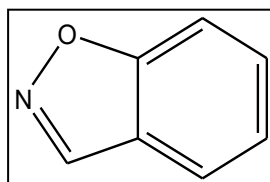


Figure 1: Structure of Benzisoxazole

In the field of Molecular modeling, docking is a computational method which estimates the preferred orientation of the ligand relative to the receptor as well as the confirmation of the ligand and receptor when bound to each other.¹⁸⁻²¹

The present research work deals with the synthesis of the title parent compounds starting from substituted 4-(benzo[d]oxazol) piperazine, followed by their molecular docking study and *in-vitro* antifungal evaluation.



METHODS AND MATERIAL:

Chemicals and Reagents

All of the chemicals were commercially available and procured from Aldrich, Merck and Vijay chemicals. All reactants and solvents were analytically pure and were used without further purification. The starting compounds, 2-bromobenzoxazole (I) and N-Boc piperazine (II) purchased from Vijay chemicals, Mehsana.

Instrumentation

Thin layer chromatography (TLC) was performed on silica gel plates (silica gel GF254) visualized at 254 nm. Infrared (IR) spectra were measured on Bruker, Alpha- II. Nuclear Magnetic resonance (NMR) spectra were measured on a Bruker biospin AV400 instrument at 400 MHz for ¹H NMR spectra. The high resolution mass spectra were acquired with a Q-TOF mass spectrometer (Impact II, Bruker).

EXPERIMENTAL:

Molecular Docking study:

Software Used: AutoDock Vina 4.2.6

Procedure: Protein-Ligand Docking

Steps for AutoDock Vina software

Firstly, choose the protein target and ligand molecule.

For protein target: use Protein Data Bank.

Process: go to PDB site-----Search required PDB-----click on download file click on PDF format ----- protein was downloading in downloads.

For ligand target: Synthetically ligand molecule.

Process: go to PubChem-----select download-----select 3D conformers save it as SDF file format ----- Ligand Was download in downloads. Both Protein and Ligand was saved in Docking Folder.

Carry out protein and ligand preparation. For Protein Preparation:

Go to AutoDock Vina tools ----- copy protein and paste in AutoDock Vina tools.

Step 1: Go to edit option -----Delete water molecules.

Step 2: Go to Edit Option ----- Go to Hydrogen-----Add hydrogen-----select on polar only ----- click ok.

Step 3: Go to Edit----- select charges----- Add Kollaman charge ----- click ok (Protein was prepared).

For PDBQT format: go to Grid ----- select Macromolecule----- click on choose----- select protein option-----

Select Molecules----- click ok ----- one tab is open--- save as PDBQT format-----ok.



For Ligand Preparation:

Step 1: Convert SDF format to Ligand PDB format

go to PyMol tool----- copy and paste ligand to PyMol tool---- go to File---- Export Molecule-----save as ligand in Docking Folder.

Step 2: Convert PDB format to PDBQT format.

Copy and paste Ligand to AutoDock Vina Tool---- go to Ligand---- select input----select Choose----- select Ligand---- select molecule for AutoDock 4----- click ok

Step 3: Save as PDBQT format.

Go to ligand---- select Output----save as PDBQT format -----Save in Docking Folder.

Select the docking site in protein.

Step 1: Preparation of Grid: copy and paste, protein and ligand in AutoDock Vina tool--- go to Grid-----Select Macromolecule----Click Choose-----Select Ligand and Protein-----Select Molecule-----Click No----- Click ok.

Step 2: For Grid Formation: go to Grid----- select Grid Box---- select Grid Option Tab open----Click on file---- choose output Grid dimension file----Tab is open -----select Docking folder----- save as grid.txt format in Docking folder.

Docking process using AutoDock Vina Tool

Installed AutoDock Vina tool----go to search--- type cmd (command prompt) ----- go to command prompt box----- "C:\Program Files (x86)\The Scripps Research Institute\Vina\vina.exe" --receptor 4dbn1.pdbqt -- ligand C36.pdbqt --config config.txt --log log.txt --out output.pdbqt -----click on enter log file and output .pdbqt file are generated in Docking folder.

Different ligand poses are generated.

Go to PyMol----copy protein----- copy output .pdbqt copy and paste.

ANTIFUNGAL ACTIVITY:

All the synthesized compounds were screened for the antifungal activity by Agar well diffusion method.

Determination of antifungal activity by Agar well diffusion method:

Composition of Media:

- Nutrient Agar: 15 %
- Peptic Digest of Animal Tissue: 5 %
- Sodium chloride: 5 %
- Bees extract: 1.5 %



- Yeast extract: 1.5 %
- Final pH (after sterilization): 7.5 ± 0.2
- Distilled water: up to 1000 ml

Micro-organism used:

- *Candida tropicalis* (Yeast)
- *Aspergillus niger* (Mould)

Standard drug used:

Fluconazole

Preparation of synthetic compounds for microbiological assay:

A stock solution of 10 mg of each synthetic compound dissolved in 1 mL of dimethyl sulfoxide (DMSO) as solvent was prepared. The antimicrobial activity of the synthesized compounds was evaluated by Agar well diffusion method.

Preparation of Standard drug for microbiological assay:

A stock solution of 1 mg of standard drug (Fluconazole) dissolved in 10 mL of dist. water as solvent was prepared.

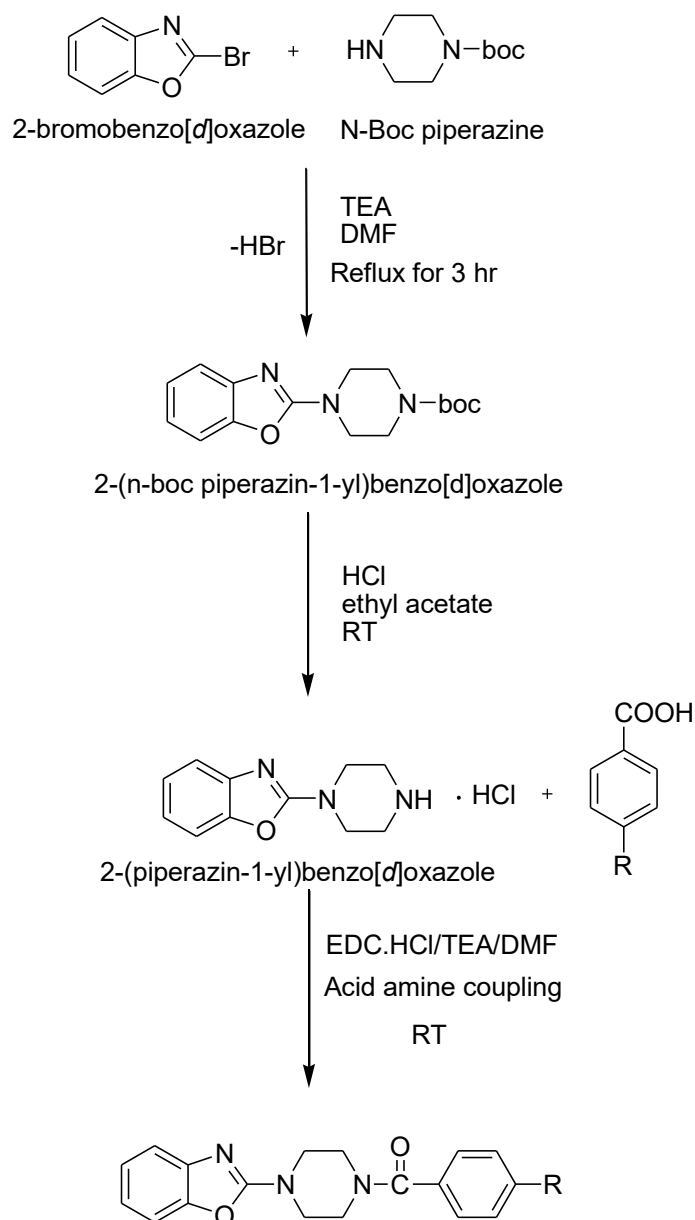
Procedure of Agar well diffusion method:

1. Requisite quantity of the liquid agar media about 15-20 ml of molten agar was spread into each sterilized Petri dish by taking the usual precautions to avoid the contaminations.
2. After solidify the liquid media test organism was spread over the solidified agar media. All the Petri dishes were marked in a specific way. Sterile cork borer was used to make well.
3. The agar plates were inoculated with the suspension of particular organism by spread plate technique.
4. Finally, the agar well plates were filled with the test solution. After the addition of the tested samples, the plates were kept in freeze for diffusion and incubated at 35°C for 24 hours for fungi in B.O.D. incubator. The zone of inhibition if any was then measured in mm for the particular compound and specific organism.
5. Standard antifungal drug Fluconazole was used as positive control and sterile distilled water as negative control and incubated at 35°C for 4- 5 days.
6. All the microbial strains used were of non-invasive species of their genera and thus applicable for analytical work. *C. tropicalis* is the fungi and *A. niger* is the mould. Their zone of inhibition has been reported.
7. The MIC was determined by measuring the absorbance of microtitre plates at 530 nm for *C. tropicalis*. While for *A. niger*, MIC was determined visually. The optical density from each well was compared with optical density from the positive control wells, the lowest concentration with optical density <0.1 signifies inhibition and considered as MIC.

RESULT AND DISCUSSION:

Chemistry

Benzimidazole derivatives were synthesized according to **Scheme 1**. According to designed scheme, six derivatives were prepared. All the intermediate and target molecules were confirmed by TLC, IR, NMR and Mass Spectra.



Scheme 1: Synthetic Scheme of Target compounds

General procedure for synthesis of newer compounds:

Synthesis of 2-(n-boc piperazin-1-yl)benzo[d]oxazole [Intermediate of STEP- 1(C-N COUPLING BY SN₂ REACTION)]:

To a stir solution of 2-bromobenzo[d]oxazole (5 gm, 1 equ) in DMF, N-Boc piperazine (1.5 equ) is added followed by drop wise addition of TEA. The reaction mixture is refluxed on the water for 3 hr and the progress of the reaction is checked by TLC. After completion of the reaction, the reaction mixture was poured into ice-cold water to obtain a precipitate. Final crude products isolated by filtration and dry it. Recrystallization was done with hot methanol.

Synthesis of 2-(piperazin-1-yl)benzo[d]oxazole hydrochloride [Intermediate of STEP- 2(Deprotection)]:

Concentrated hydrochloric acid (5 mL) was added to N-Boc piperazinyl benzo[d]oxazole (step 1 product) (4.0 g). The resulting solution was stirred for 2 h and Ethyl acetate (20 mL) was then added over a period of 1 h at room temperature. The progress of the reaction was checked by TLC and after completion of the reaction, the resulting slurry was filtered. The filter cake was dried and then dissolved in ethyl acetate (20 mL) and H₂O (10 mL). The crude product was isolated by extraction with 2 gm.

**Synthesis of substituted benzisoxazole (Acid-Amine coupling):**

Take solution of 2-piperazinyl benzoxazole (1.2 equ) in DMF in a round-bottomed flask. EDC HCl (1 equ) and TEA (1.5 equ) were added and the reaction mixture was stirred at RT for about 5-10 min. After that benzoic acid derivatives (1.2 equ) were added and the reaction mixture was stirred at room temperature for 3 hours. After the completion of the reaction monitored by TLC, the reaction mixture was diluted with water and extracted with DCM, dried in sodium sulphate and distilled in reduced pressure to obtain the final product.

Physical and Spectral characterization of newer compounds:**Table 1: Physical characterization data of Intermediates**

Intermediates	Molecular Formula	IUPAC name	Mol. Wt. (g/mol)	Melting Point (°C)	Yield %	*R _f
Step- 1 Intermediate	C ₁₆ H ₁₄ N ₂ O ₂	2-(n-boc piperazin-1-yl)benzo[d]oxazole	420.49	88-90	90.21	0.2
Step- 2 Intermediate	C ₁₁ H ₁₄ ClN ₃ O	2-(piperazin-1-yl)benzo[d]oxazole hydrochloride	239.08	172-175	45.62	0.4

Mobile phase combination used for TLC:*Hexane: Ethyl acetate (9:1)

Table 2: Physical characterization data of synthesized compounds

Compound	R	Molecular formula	IUPAC name	Mol. Wt. (g/mol)	Melting Point (°C)	Yield %	*R _f
1	-H	C ₁₈ H ₁₇ N ₃ O ₂	(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(phenyl)methanone	307.35	165-168	80.0	0.8
2	-NO ₂	C ₁₈ H ₁₆ N ₄ O ₄	(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(4-nitrophenyl)methanone	352.34	184-187	79.0	0.8
3	-CH ₃	C ₁₉ H ₁₉ N ₃ O ₂	(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(p-tolyl)methanone	321.37	154-157	83.0	0.7
4	-Cl	C ₁₈ H ₁₆ ClN ₃ O ₂	(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(4-chlorophenyl)methanone	341.79	178-182	67.0	0.75
5	-OCH ₃	C ₁₉ H ₁₉ N ₃ O ₃	(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(4-methoxyphenyl)methanone	337.37	138-142	82.0	0.8
6	-Br	C ₁₈ H ₁₆ BrN ₃ O ₂	(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(4-bromophenyl)methanone	386.24	170-173	45.0	0.7

Mobile phase combination used for TLC:*Chloroform: methanol (9.5:0.5)

Spectral characteristics:**(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(phenyl)methanone**

IR in cm⁻¹ (KBr) Aromatic CH: 3063, 3033, C=O (Amide): 1689, C=N: 1607, C-O:1195, C-N:1461;

¹H NMR in δ ppm (DMSO, 400 MHz) 3.36-3.43 (-CH₂, Methylene), 7.44-7.95 (Aromatic- H, 1-benzene), 7.26 (Aromatic- H,



Benzene);

Mass M⁺ peak (m/z) 307.35 (M+H)⁺

(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(4-nitrophenyl)methanone

IR in cm⁻¹ (KBr) Aromatic CH: 3077 C=O: 1688, C=N: 1608, C-O:1084, NO₂ : 1577, 1462, C-N:1310;

¹H NMR in δ ppm (DMSO, 400 MHz) 3.36-3.43 (-CH₂, Methylene), 8.21-8.37 (Aromatic- H, 1-benzene), 7.26 (Aromatic- H, Benzene);

Mass M⁺ peak (m/z) 386.24 (M+H)⁺

(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(p-tolyl)methanone

IR in cm⁻¹ (KBr) Aromatic CH: 3059, 3031, 3007, Aliphatic CH: 2923, 2871, C=O (Amide): 1688, C=N: 1607, C-O:1025, C-N:1461;

¹H NMR in δ ppm (DMSO, 400 MHz) 3.36-3.43 (-CH₂, Methylene), 7.24-7.83 (Aromatic- H, 1-benzene), 7.26 (Aromatic- H, Benzene), 2.35 (-CH₃, Methyl);

Mass M⁺ peak (m/z) 341.79 (M+H)⁺

(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(4-chlorophenyl)methanone

IR in cm⁻¹ (KBr) Aromatic CH: 3069, C=O (Amide): 1679, C=N: 1609, C-O:1116, C-N:1460;

¹H NMR in δ ppm (DMSO, 400 MHz) 3.36-3.43 (-CH₂, Methylene), 7.45-7.89 (Aromatic- H, 1-benzene), 7.26 (Aromatic- H, Benzene);

Mass M⁺ peak (m/z) 352.34 (M+H)⁺

(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(4-methoxyphenyl)methanone

IR in cm⁻¹ (KBr) Aromatic CH: 3077, Aliphatic CH: 2974, C=O (Amide): 1690, C=N: 1608, C-O:1116, C-N:1409;

¹H NMR in δ ppm (DMSO, 400 MHz) 3.36-3.43 (-CH₂, Methylene), 6.95-7.84 (Aromatic- H, 1-benzene), 7.26 (Aromatic- H, Benzene), 3.73 (-CH₃, Methyl);

Mass M⁺ peak (m/z) 321.37 (M+H)⁺

(4-(benzo[d]oxazol-2-yl)piperazin-1-yl)(4-bromophenyl)methanone

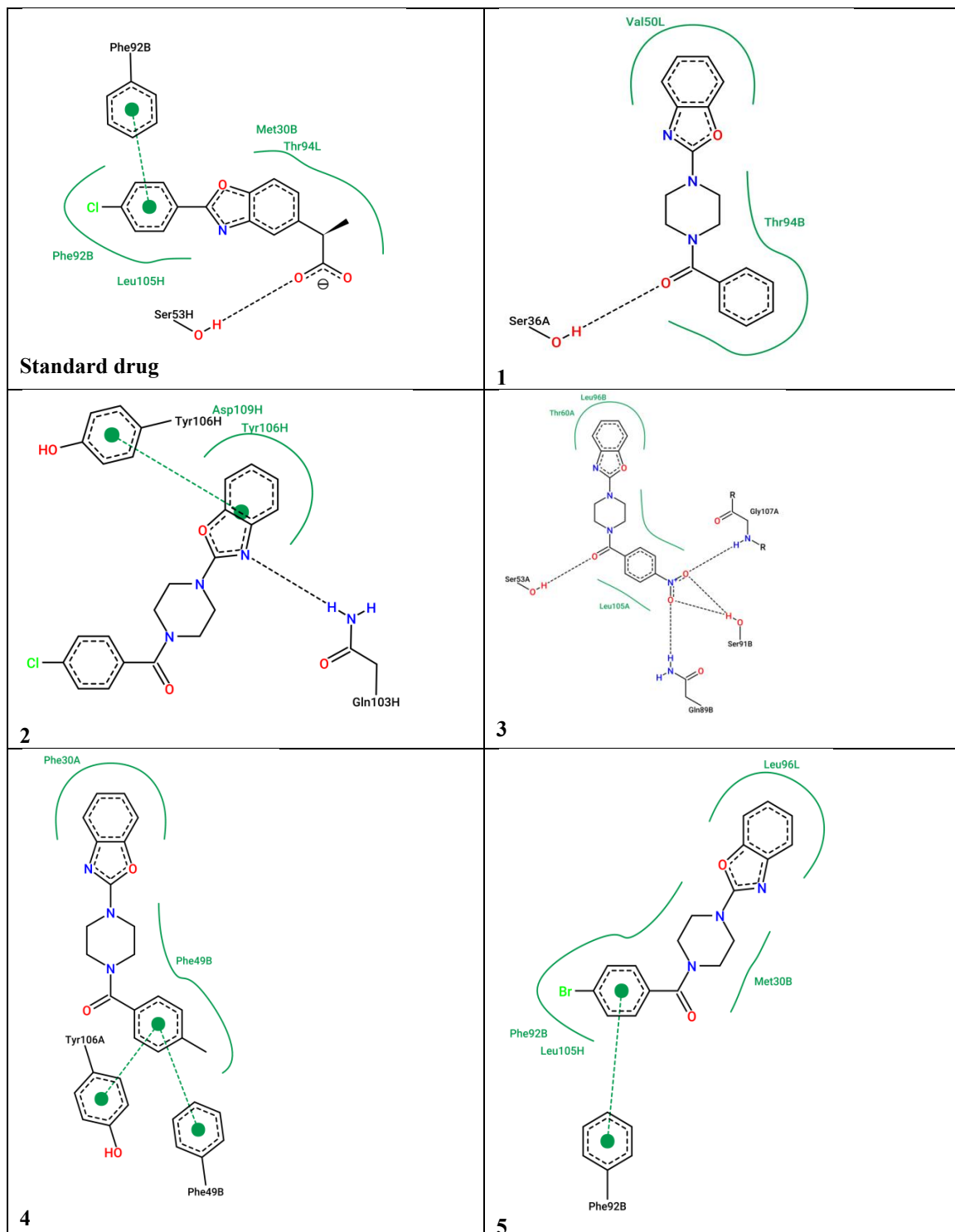
IR in cm⁻¹ (KBr) Aromatic CH: 3077, C=O (Amide): 1690, C=N: 1609, C-O:1121, C-N:1462;

¹H NMR in δ ppm (DMSO, 400 MHz) 3.36-3.43 (-CH₂, Methylene), 7.61-7.84 (Aromatic- H, 1-benzene), 7.26 (Aromatic- H, Benzene);

Mass M⁺ peak (m/z) 337.37 (M+H)⁺



Docking score



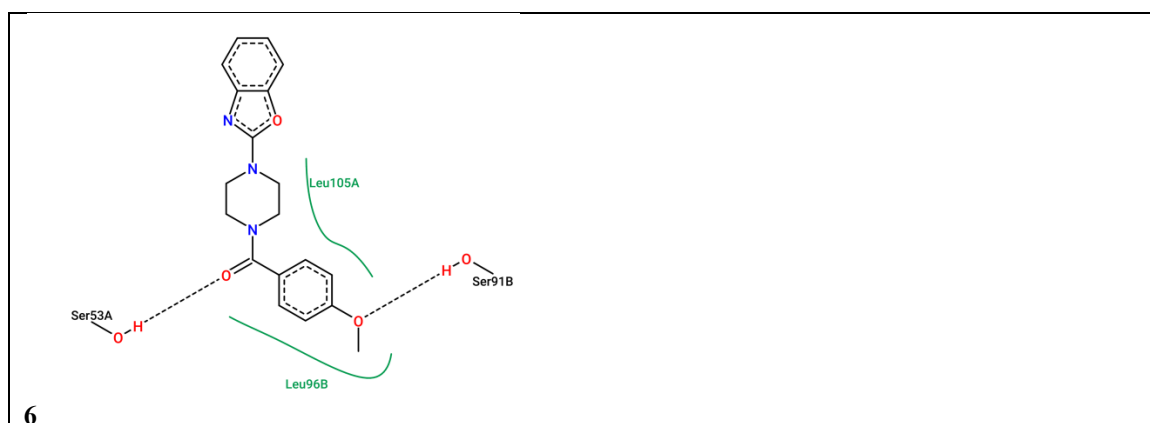


Figure 2: 2D interaction image Molecular docking study of derivatives

Table 3: Docking score of benzisoxazole derivatives

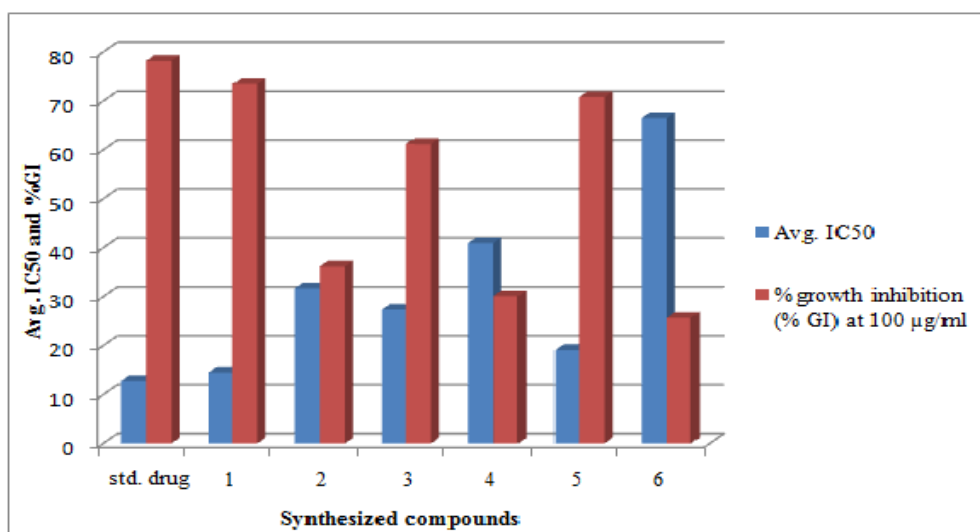
Compounds	Docking Binding energy (kcal/mol)
Fluconazole	-9.7
1 (Phenyl)	-8.3
2 (p-nitro)	-9.3
3 (p-methyl)	-8.4
4 (p-chloro)	-8.6
5 (p-methoxy)	-8.7
6 (p-bromo)	-8.0

From the docking study, we conclude that compound 2, compound 4 and compound 5 may give anti fungal activity. Among all these compounds, **compound 2 may give better antifungal activity.**

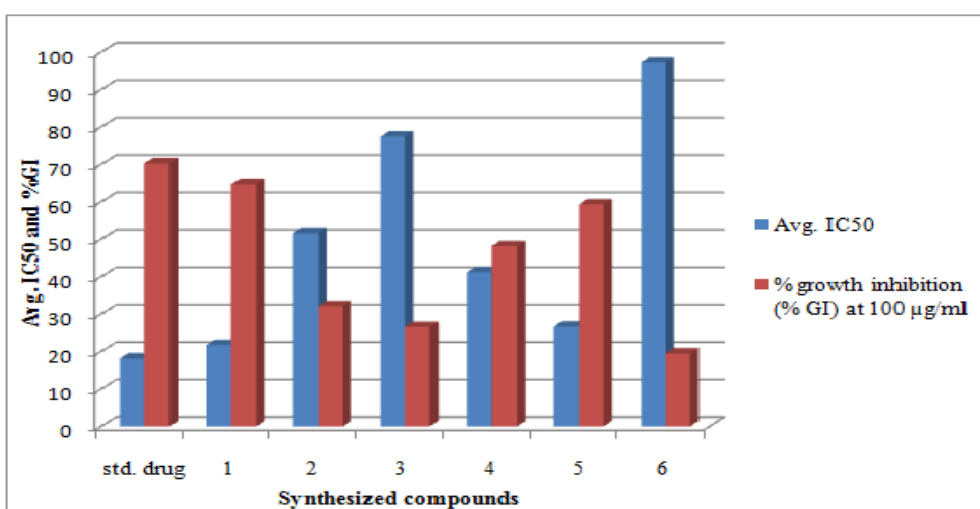
Antifungal activity:

Table 4: Comparison of IC 50 of synthesized Compounds with standard drug

Compounds	<i>A. niger</i> (715)		<i>C. tropicalis</i> (2168)	
	Avg. IC50	% growth inhibition (% GI) at 100 µg/ml	Avg. IC50	% growth inhibition (% GI) at 100 µg/ml
1	14.48	73.5	21.71	64.7
2	31.67	36.2	51.64	32.1
3	27.36	61.2	77.51	26.6
4	40.95	30.1	41.19	48.2
5	19.11	70.8	26.71	59.4
6	66.48	25.7	97.30	19.5
Std. Drug (Fluconazole)	12.74	78.2	18.21	70.3



Graph 1: Comparison of avg. IC50 and %GI against *A. niger* of all compounds with standard drug



Graph 2: Comparison of avg. IC50 and %GI against *C. tropicalis* of all compounds with standard drug

CONCLUSION

In this study, we designed, synthesized the series of 4-(benzo[d]oxazol) piperazine derivatives. All compounds are successfully characterized with TLC, IR, NMR, Mass Spectra and screened for their chemotherapeutics anti fungal activity. From the graph I and II, It concluded that compound 1 and compound 5 gives better activity against *A. niger* (715) and *C. tropicalis* (2168). Whereas **compound 1 gives better activity** then compound 5 because it contains phenyl group giving stable compound compare to compound 5 contains methoxy group at p- position.

CONFLICT OF INTEREST:

The authors declare no conflict of interest.

ACKNOWLEDGEMENT:

We express sincere thanks to Faculty of Pharmacy, SRI campus for providing laboratory facilities for our research work.



REFERENCES:

1. Gringauz A. (1997) Introduction to medicinal chemistry: how drugs act and why. Wiley-VCH Inc., USA, pp. 295-296.
2. Finegold SM, Baron EJ. (1986) Bailey and Scott's Diagnostic Microbiology. CV Mosby, St. Louis, pp. 678-774.
3. Andriole VT, "Current and future antifungal therapy: new targets for antifungal agents." *J. Antimicro. Chemoth.*, **1999**, *44*, 151-162.
4. Georgopapadakou NH, Walsh TJ, "Antifungal agents: chemotherapeutic targets and immunologic strategies." *Antimicro. agents chemoth.*, **1996**, *40*, 279- 291.
5. Selitrennikoff CP, "Antifungal proteins." *App. Environ. Micro.*, **2001**, *67*, 2883-2894.
6. Katritzky AR, Pozharskii AF. (2000) Handbook of Heterocyclic Chemistry, 2nd Edn., Academic Press.
7. Clayden J, Greeves N, Warren S, Wothers P. (2001) Organic Chemistry. Oxford, Oxfordshire: Oxford University Press.
8. Domene C, Jenneskens LW, Fowler PW, "Aromaticity of anthranil and its isomers, 1,2-benzisoxazole and benzoxazoles." *Tetrahedron Lett.*, **2005**, *46*(23), 4077-4080.
9. Kemp DS, Woodward RB, "The N-ethylbenzisoxazolium cation—I." *Tetrahedron*, **1965**, *21*(11), 3019-3035.
10. Casey ML, Kemp DS, Paul KG, Cox DD, "Physical organic chemistry of benzisoxazoles. I. Mechanism of the base-catalyzed decomposition of benzisoxazoles." *J. Org. Chem.*, **1973**, *38*(13), 2294-2301.
11. Kemp DS, Cox DD, Paul KG, "Physical organic chemistry of benzisoxazoles. IV. Origins and catalytic nature of the solvent rate acceleration for the decarboxylation of 3-carboxybenzisoxazoles." *J American Chem. Soc.*, **1975**, *97*(25), 7312-7318.
12. Pandit Y, Sahani R, Liu R, "Gold-Catalyzed Michael-Type Reactions and [4 + 2]-Annulations between Propiolates and 1,2-Benzisoxazoles with Ester-Directed Chemoselectivity." *Org. Lett.*, **2018**, *20*(21), 6655-6658.
13. Kaur A, Wakode S, Pathak D, Sharma P, "Synthesis and Biological Evaluation of Some New 5-Acylamino-2-(3,4,5-Trimethoxyphenyl)Benzoxazoles as Potential Antibacterial and Anti-inflammatory Agents." *Ind. J. Heterocyc. Comp.*, **2017**, *27*(4), 347-354.
14. Kakkar S, Kumar S, Narasimhan B, Lim S, Ramasamy K, Mani V, Shah S, "Design, synthesis and biological potential of heterocyclic benzoxazole scaffolds as promising antimicrobial and anticancer agents." *Chem. Central J.*, **2018**, *12*, 96.
15. Rodrigues M, Sharath BS, Bennehalli B, Vagdevi HM, "Synthesis, biological evaluation and molecular docking studies of heterocycle encompassed benzoxazole derivatives as antimicrobial agents." *Mat. Todays: Proceedings*, **2021**, *49*(3), 849-853.
16. Luo B, Li D, Zhang AL, Gao JM, "Synthesis, Antifungal Activities and Molecular Docking Studies of Benzoxazole and Benzothiazole Derivatives." *Molecules*. **2018**, *23*(10), 2457.
17. Carvalho D, Alvarenga D, Carmo L, Oliveira L, Silva N, Dias A, Coelho L, Dias D, "Antifungal Activity of New Eugenol-Benzoxazole Hybrids against *Candida* Spp." *J. Chem.*, **2017**, *6*, 1-8.
18. Gohlke H, Hendlich M, Klebe G. Knowledge-based scoring function to predict protein-ligand interactions. *Journal of Molecular Biology*, 2000; 295: 337-356. <https://doi.org/10.1006/jmbi.1999.3371>
19. Thomsen R, Christensen MH. MolDock: A new technique for high-accuracy molecular docking. *Journal of Medicinal Chemistry*, 2006; 49: 3315-3321. <https://doi.org/10.1021/jm051197e>
20. Tietze S, Apostolakis J. GlamDock. Development and validation of a new docking tool on several thousand protein-ligand complexes. *Journal of Chemical Information and Modelling*. 2007; 47(4): 1657- 1672. <https://doi.org/10.1021/ci7001236>
21. Fan H, Schneidman-Duhovny D, Irwin JJ, Dong G, Shoichet BK, Sali A. Statistical potential for modelling and ranking of protein-ligand interactions. *Journal of Chemical Information and Modelling*. 2011; 51(12): 3078-3092. <https://doi.org/10.1021/ci200377u>

How to cite this article:

Dr. Ojas Patel et al. *Ijppr.Human*, 2026; Vol. 32 (4): 95-105

Conflict of Interest Statement: All authors have nothing else to disclose.

This is an open access article under the terms of the Creative Commons Attribution-NonCommercial-NoDerivs License, which permits use and distribution in any medium, provided the original work is properly cited, the use is non-commercial and no modifications or adaptations are made.